31190

S/079/61/031/012/004/011 D228/D301

The aryl trifluoromonohydrides ...

conversions. Thus, II was almost completely converted into n-chlorophenyl phosphinic acid on the addition of water, while n-chlorophenyl thiophosphinic acid (III) was obtained with a 96% yield by heating II and S to a temperature of 140-150°. The chlorination of I gave mixed phosphorous tetrachlorofluorides which were identified by means of their hydrolysis to the corresponding phenyl phosphinic acids (yield 87%). The authors give details for synthesizing three fluoroanhydrides of n-chlorophenyl monothiophosphinate from III, alcohol and Et<sub>3</sub>N:n-clc<sub>6</sub>H<sub>4</sub>PS(OMe)F - yield 48%; n-clc<sub>6</sub>H<sub>4</sub>PS(OEt)F - yield 80%; and n-clc<sub>6</sub>H<sub>4</sub>PS(Oiso-Pr)F - yield 65%. These fluoroanhydrides which do not corrode glass in the absence of moisture and which possess active contact-insecticide properties, are readily soluble in standard organic solvents, insoluble in water, and slow-ly hydrolyzed at 20°. There are 2 tables and 8 references: 4 Soviet-bloc and 4 non-Soviet-bloc. The references to the English-language publications read as follows: J. Van Wazer et al, J. Amer. Chem. Soc. 78, 5715, (1956); A. Burg et al, ibid, 80, 3198 (1958):

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The aryl trifluoromonohydrides ...

S/079/61/031/012/004/011 D228/D301

and G. Kosolapoff, Organophosphorous compounds; 147, (1950).

ASSOCIATION:

Institut organicheskoy khimii, Akademii nauk Ukrain-skoy SSR (Institute of Organic Chemistry, Academy of Sciences, Ukrainian SSR)

SUBMITTED:

January 6, 1961

Card 3/3

KIRSANOV, A.V. [Kirsanov, O.V.], akademik; SHTEPANEK, A.S.; SHEVCHENKO, V.I.

Reaction of diphenyl phosphorus trichloride and diphenyl phosphorus dichloride with urethan. Dop. AN URSR no.1:63-65 162. (MIRA 15:2)

1. Institut organicheskoy khimii AN USSR. 2. AN USSR (for Kirsanov.

(Phosphorus organic compounds)
(Carbamic acid)

SHEVCHENKO, V.I.; SHTEPANEK, A.S.; KIRSANOV, A.V.

3-Arylcarbamidophenylphosphinic acids and their chlorides. Zhur. ob khim. 32 no.1:150-153 Ja '62. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphinic chloride)

DERKACH, G.I.; DREGVAL', G.F.; KIRSANOV, A.V.

Trianilidophosphazo-N-arylsulfonylaminobenzoyls and M'-dianilidophosphinyl-N'-arylsulfonylbenzamidines. Zhur. ob. khim. 32 no.1:154-159 Ja '62. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphorus organic compounds)

DERKACH, G.I.; SHOKOL, V.A.; SAMARAY, L.I.; KIRSANOV, A.V.

New method of preparing trichlorophosphazoacyls. Zhur. ob. khim. 32 no.1:159-160 Ja 162. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphazo compounds)

LEVCHENKO, Ye.S.; KIRSANOV, A.V.

Sulfur N-arylsulfonyliminodioxide. Zhur. ob. khim. 32 no.1:161-165 Ja 162. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR.
(Sulfur dioxide) (Sulfur organic compounds)

SHOKOL, V.A.; DERKACH, G.I.; KIRSANOV, A.V.

Phenyldichloro- and diphenylchlorophosphazo-dichloro-and trichloroacetyls and their derivatives. Zhur. ob. khim. 32 no.1: 166-171 Ja '62. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphaso compounds)

DERKACH, G.I.; LEPESA, A.M.; KIRSANOV, A.V.

Alkyl esters of N-dialkoxyphosphinyliminocarboxylic acids.

Zhur. ob. khim. 32 no.1:171-174 Ja 162. (MIRA 15:2)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphinic acid)

KIRSANOV, A.V., [Kirsanov, O.V.] akademik; LEPESA, A.M.; DERKACH, G.I. [Derkach, H.I.] Et ers of monoanilides of aroylamidophosphoric acids. Dop. AN (MIRA 15:5) URSR no.3:384-386 162.

1. Institut organicheskoy khimii AN USSR. 2. AN USSR (for Kirsanov).

(Phosphoramidic acid) (Ethers)

DERKACH, G.I.; GUBNITSKAYA, Ye.S.; SHOKOL, V.A.; KIRSANOV, A.V.

Triaroxyphosphazoacyls. Part 2. Zhur.ob.khim. 32 no.4:1201-1207 Ap '62. (MIRA 15'4)

 Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphorus organic compounds) (Esters)

LEVCHENKO, Ye.S.; DERKACH, N.Ya.; KIRSANOV, A.V.

N-arylsulfonylareniminosulfenamides. Zhur.ob.khim. 32 no.4: 1208-1212 Ap '62. (MIRA 15'4)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Sulfonsmides)

LEVCHENKO, Ye.S.; DERKACH, N.YA.; KIRSANOV, A.V.

Reaction of diaryldisulfonyl imides with phosphorus pentachioride. Zhur.ob.khim. 32 no.4:1212-1218 Ap '62. (MIRA 15:4)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Sulfonyl group) (Phosphorus chlorides)

DERKACH, G.I.; GUBNITSKAYA, Ye.S.; SHOKOL, V.A.; KIRSANOV, A.V.

Phenyldichloro-, diphenylchloro-, and triphenylphosphazo acyls. Zhur.ob.khim. 32 no.6:1874-1878 Je 162. (MIRA 15:6)

1. Institut organicheskoy khimii Akadamii nauk Ukrainskoy SSR. (Phosphaso compounds)

DERKACH, G.I.; DREGVAL, G.F.; KIRSANOV, A.V.

Derivatives of phosphorylated amidines. Zhur.ob.khim. 32 no.6:1878-1882 Je \*62. (MIRA 15:6)

1. Institut organicheskoy khimii Akademii nauk Ukrainskoy SSR. (Amidine) (Phosphorus organic compounds)

DERKACH, G.I.; KIRSANOV, A.V.

Phosphorylated amidines. Zhur.ob.khim. 32 no.7:2254-2256 Jl \*62.
1. Institut organicheskoy khimii AN USSR.
(Amidines) (Phosphorous acid)

LEVCHENKO, Yo.S.; KIRSANOV, A.V.

Sulfur bisarylsulfonyliminodioxide. Zhur.ob.khim. 32 no.7:2256-2262 Jl 162. (HIRA 15:7)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Sulfur organic compounds)

LEVCHENKO, YE.S., PIVEN, YU.V., KIRSANOV, A.V.

Reaction of phosphorus diiodide with alkyl halides.

Khimiya i Primetaniye Fosfororganichaskikh Soyadin niv (Chamietry and application of organophosphorus compounds) A. YE. 8:4 207, 64, 400, 600, by Kazar Affil. Acad. 161. USSR, Moscow 1962, 1987 pp.

Collection of complete papers presented at the 1909 Kazan Southerence on Chemistry of Tryanophosphorus Compounds.

DERKACH, G.I.; ZHURAVLEVA, L.P.; KIRSANOV, A.V.

N-dichlorophosphinyl-N'-aryl-C-chloroformamidines. Zhur.ob.khim. 32 no.3:879-881 Mr '62. (MIRA 15:3)

1. Institut organicheskoy khimii AN Ukrainskoy SSR.
(Formamidine) (Phosphinic chloride)

LEVCHENKO, Ye.S.; KOZLOV, E.S.; KIRSANOV, A.V.

Esters of N-carbethoxyareneiminosulfonic acids. Zhur.ob.khim. 32 no.3:882-886 Mr '62. (MIRA 15:3)

1. Institut organichaskoy khimii AN Ukrainskoy SSR. (Sulfonic acids)

KIRSANOV, A.V.; KIRSANOVA, N.A.

N-arylaulfonylathylene- and hexamethylanediamines. Zhur.ob.khim, 32 no.3:887-892 Mr '62. (MIRA 15:3)

1. Institut organicheskoy khimii i Institut monomerov i polimerov AN Ukrainskoy SSR.

(Ethylenediamine) (Hexanediamine)

BELYAKOVA, L.D.; KISELEV, A.V.

Adsorption of nonpolar molecules having different electronic shell structure on the adsorbents of different nature. Report No.1: Adsorption of benzene and n.haxane on barium sulfate. Izv.AN SSSR.Otd.khim.nauk no.6:969-974 '62. (MIRA 15:8)

1. Institut fizicheskoy khimii AN SSSR i Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova.

(Benzene) (Hexane) (Adsorption)

DZHIGIT, O.M.; ZHDANOV, S.P.; KISELEV, A.V.; MUTTIK, G.G.

Differential heats of adsorption of n-pentane and diethyl ether by porous crystals of seclite of type 5A. Zhur. fis. khim. 36 no.4:919-920 Ap '62. (MIRA 15:6)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, khimicheskiy fakul'tet i Institut khimii silikatov AN SSSR. (Heat of adsorption) (Pentane) (Ethyl ether) (Zeolite crystals)

ZHDANOV, S.P.; KALMANOVSKIY, V.I.; KISELEV, A.V.; FIKS, M.M.; YASHIN, Ya.I.

Use of porous glasses as adsorbents in gas chromatography.

Zhur.fiz.khim. 36 no.5:1118-1120 My '62. (MIRA 15:8)

1. Institut khimii silikatov AN SSSR; Opytno-konstruktorskoye byuro avtomatiki Gosudarstvennogo komiteta khimicheskoy promyshlennosti pri Sovete Ministrov SSSR, Dzershinskiy filial i Moskovskiy gosudarstvennyy universitet imeni Lomonosova, khimicheskiy fakul'tet.

(Glass) (Adsorbents) (Gas chromatography)

AKSHINSKAYA, N.V.; KISELEY, A.V.; MIKITIN, Yu.S.; PETROVA, R.S.; CHUYKINA, V.K.; SHCHERBAKOVA, K.D.

Geometric and chemical modification of silica gel for the adsorption separation of hydrocarbons by gas chromatography. Zhur.fiz.khim. 36 no.5:1121-1123 My '62. (MIRA 15:8)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.
(Silica) (Hydrocarbons) (Gas chromatography)

ZHMUROVA, I.N.; KIRSANOV, A.V.

Mechanism of phosphaso reaction. Zhur.ob.khim. 32 no.8:2576-2580 Ag 162. (MIRA 15:9)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphaso compounds)

IVANOVA, Zh.M.; KIRSANOV, A.V.

Phosphorus aryltrifluoromonochlorides and N,N-disubstituted phosphorus aryltrifluoromides. Zhur.ob.khim. 32 no.8:2592-2595 Ag '62. (MIRA 15:9)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphorus organic compounds)

## ZHMUROVA, I.N.; KISILENKO, A.A.; KIRSANOV, A.V.

Infrared spectra of monomer and dimer trichlorophosphase aryls and phenyldichlorophosphase aryls. Zhur. Sp. khim. 32 no.8:2580-2585 Ag 162. (MIRA 15:9)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Phosphazo compounds--Spectra)

LEVCHENKO, Ye.S.; KOZLOV, E.S.; KIRSANOV, A.V.

Phenyl esters of areminino sulfonic avids. Zhur.ob.khim. 32 no.8:2585-2592 Ag '62. (MIRA 15:9)

1. Institut organicheskoy khimii AN Ukrainskoy SSR. (Eulfonic soid) (Esters)

SHEVCHENKO, V.I.; SHTEPANEK, A.S.; KIRSANOV, A.V.

Esters of diphenylchloro- and triphenylphosphazo carbonic acid. Zhur.ob.khim. 32 no. 1:2595-2600 Ag 162. (MIRA 15:9)

1. Institut organic eskoy khimii AN Ukrainskoy SSR. (Phosphazo c mpounds) (Carbonic acid) (Esters)

DERKACH, G.I.; LEPESA, A.M.; KIRSANOV, A.V.

Derivatives of monoanilides of aroylamidophosphoric acids. Zhur.ob.khim. 32 no.8:2600-2606 Ag 162. (MIRA 15:9)

1. Institit organicheskoy khimii AN Ukrainskoy SSR. (Phosphoramidic acid)

DERKACH, G.I.; PROTSENKO, L.D.; ZHURAVLEVA, L.P.; KIRSANOV, A.V.

N-diethylenediamidophosphinyl-N'acthylene-N"-arylguanidines. Zhur.ob.khim. 32 no.9:2992-2994 S '62. (MIRA 15:9)

1. Institut organicheskoy khimii AN UkrSSR. (Guanidine) (Phosphorylation)

SHEVCHENKO, V.I.; BODNARCHUK, N.D.; KIRSANOV, A.V.

Phosphorylation of malonic c.d esters. Zhur.ob.khim. 32 no.9:2994-3001 S 162. (MIRA 15:9)

1. Institut organicheskoy khimii AN UkrSSR.
(Malonic acid) (Phosphorylation)

DERKACH, G.I.; DREGVAL!, G.F.; KIRSANOV, A.V.

N-dichlorophosphinyl-N'd,-chlorobenzalarenamidines. Zhur.ob.khim. 32 no.9:3002-3007 S '62. (MIRA 15:9)

1. Institut organichaskoy khimii AN UkrSSR.
(Benzamidine) (Phosphorus organic compounds)

KIRSANOV, A.V., akademik; LOBOV, V.P., kand.biolog.nauk

Effective control measures against the sugar beet weevil. Vest. AN SSSR 32 no.11:95-96 N '62. (MIRA 15:11)

1. AN UkrSSR (for Kirsanov).
(Ukraine—Sugar beets—Diseases and pests)
(Ukraine—Weevils) (Insecticides)

ZHURAVLEVA, L.P.; KIRSANOV, A.V.

Phosphorylation of aminoarenesulfonamides. Zhur.ob.khim. 32 no.11:3752-3 N 162. (MIRA 15:11)

1. Institut organicheskoy khimii AN UkrSSR. (Sulfonamides) (Phosphorylation)

DERKACH, G.I.; SAMARAY, L.I.; SHTEPANEK, A.S.; KIRSANOV, A.V.

Alkyl esters of phosphasocarbonic acid. Zhur.ob.khim. 32 no.11:3759-3761 N '62. (MIRA 15:11)

 Institut organicheskoý khimii AN UkrSSR. (Phosphazo compounds) (Carbonic acid)

DERKACH, G.I.; SAMARAY, L.I.; KIRSANOV, A.V.

Reaction of iminoesters with phosphorus pentachloride. Zhur.ob.khim. 32 no.11:3761-3764 N '62. (MIRA 15:11)

 Institut organicheskoy khimii AN UkrSSR. (Esters)
 (Phosphorus chloride)

IVASHCHENKO, Ya.N.; MOSHCHITSKIY, S.D.; KIRSANOV, A.V.

Alkyl aryl esters of oxalic acid. Zhur.ob.khim. 32
no.11:3765-3768 N '62. (MIRA 15.11)

(Oxalic acid)

SHEVCHENKO, V. I.; TKACH, V. P.; KIRSANOV, A. V.

Triallylhydroxyphosphaso sulfonyl aryls. Zhur. ob. khim. 32 no.12:4047-4049 D \*62. (MIRA 16:1)

1. Institut organichaskoy khimii AN Ukrainskoy SSR.

(Phosphaso compounds) (Sulfonyl group)

\$/079/63/033/001/010/023 D205/D307

AUTHORS:

Zhmurova, I. N. and Kirsanov, A. V.

TITLE:

The acidolysis of monomeric and dimeric phenyldichlo-

rophosphazoaryls

PERIODICAL: Zhurnal obshchey khimii, v. 33, no. 1, 1963, 182-188

Compounds  $C_6H_5P(0)(NHAr)C1$  (I), where  $Ar=C_6H_5$ ,  $\underline{m}-CH_3\cdot C_6H_4$ , TEXT: p-CH3.C6H4, m-ClC6H4, pCH3OC6H4 and p-EtOC6H4 were prepared by monomerizing (ArN=PCl2C6H5)2 by boiling with benzene, cooling the monomeric solution and treating it with acetic acid. The reactions could also be carried out without isolating the dimers prior to monomerization. Compounds of type I, where Ar=o-CH3C6H4, o-ClC6H4,  $\underline{\mathbf{m}}$ -ClC<sub>6</sub>H<sub>4</sub>, 2,4-Cl<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, 2,4,6-Cl<sub>3</sub>C<sub>6</sub>H<sub>2</sub>,  $\underline{\mathbf{o}}$ -BrC<sub>6</sub>H<sub>4</sub>,  $\underline{\mathbf{m}}$ -BrC<sub>6</sub>H<sub>4</sub>,  $\underline{\mathbf{p}}$ -BrC<sub>6</sub>H<sub>4</sub>, 2,4-Br<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, 2,4,6-Br<sub>3</sub>C<sub>6</sub>H<sub>2</sub>, o-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, m-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>, p-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>,

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CIA-RDP86-00513R000722720006-5" **APPROVED FOR RELEASE: 06/13/2000** 

The acidolysis of ...

S/079/63/033/001/010/023 D205/D307

 $(NO_2)_2C_6H_3$  and 2,6- $Cl_2$ -4- $NO_2C_6H_2$  were also made, by the acidolysis of monomeric phenyldichlorophosphazoaryls with CH3COOH, using benzene or CCl, as solvent. The yields of compounds I varied between 57 and 97%. Polymeric anhydroaryliminophenylphosphinic acids [ArnP(0)C6H5]n, where Ar=C6H5, p-CH3C6H4, p-CH OC6H4, and p-C2H5OC6H4 were obtained by the acidolysis of (ArN=PCl2C6 15)2 without monomerization, with gentle heating over 5 - 6 hrs together with CH3COOH in benzene solution, in 52 - 87% yields. There is 1 table.

ASSOCIATION:

Institut organicheskoy khimii Akademii nauk Ukrainskoy SSR (Institute of Organic Chemistry of the Aca-

demy of Sciences of the Ukrainian SSR

SUBMITTED:

January 10, 1962

Card 2/2

SHEVCHENKO, V.I.; TKACH, V.P.; KIRSANOV, A.V.

Diallyl esters of arylsulfonyismidophosphoric acids. Zhur.ob. khim. 33 no.2:562-564 F \*163. (MIRA 16:2)

1. Institut organicheskoy khimii AN UkrSSR.
(Phosphoramidic acid)

LEVCHENKO, Ye.S.; KOZLOV, E.S.; KIRSANOV, A.V.

Amides of areniminosulfonic acids. Zhur.ob.khim. 33 no.2:565-571 F 163. (MIRA 16:2)

1. Institut organicheskoy khimii AN UkrSSR. (Sulfonamides)

FEDOROVA, G.K.; KIRSANOV, A.V.

Styryldialkyl phospines and th ir oxides. Zhur.ob.khim. 33 no.3:1011-1013 Mr 163. (MIRA 16:3)

1. Institut organicheskoy khimii AN UkrSSR.
(Phosphine)
(Phosphine oxide)

FESHCHENKO, N.G.; ALEKSEYEVA, T.I.; KIRSANOV, A.V.

Alkylation of phosphorus diiodite with higher alkyl iodides. Zhur.ob.khim. 33 no.3:1013-1014 Mr '63. (MIRA 16:3)

1. Institut organicheskoy khimii AN UkrSSR.
(Phosphorus iodides)
(Alkyl iodides)

ZHMUROVA, I.N.; KIRSANOV, A.V.

Diphenylchlorophosphaso aryls. Zhur.ob.khim. 33 no.3:1015-1017 Mr 163. (MIRA 16:3)

1. Institut organicheskoy khimii AN UkrSSR. (Phosphorus organic compounds)

CIA-RDP86-00513R000722720006-5

SHEVCHENKO, V.I.; BONDARCHUK, N.D.; MRRSANOV, A.V.

Trichlorophosphazoperchlorovinyl and trichlorophosphazoperchloroethyl. Zhur.ob.khim. 33 no.4:1342-1345 Ap 163. (MIRA 1615)

1. Institut organicheskiy khimii AN UkrSSR. (Phosphazo compounds)

LEVCHENKO, Ye.S.; BAL'ON, Ya.G.; KIRSANOV, A.V.

Condensation of sulfur M-aryl sulfunyimonomia.

dienes. Zhur. ob, khim. 33 no.5:1579-1584 My 163.

(MIRA 16:6) Condensation of sulfur M-aryl sulfonylmonoiminodioxides with

1. Institut organicheskoy khimii AN UkrSSR.
(Sulfonyl group) (Butadiene)
(Condensation products)Chemistry))

DERKACH, G.I.; NARBUT, A.V.; KIRSANOV, A.V.

Reaction of phosphorus pentachloride with aryl ureas. Zhur. ob. khim. 33 no.5:1584-1587 My '63. (MIRA 16:6)

1. Institut organicheskoy khimii AN UkrSSR. (Phosphorus chlorides) (Urea)

SHEVCHENKO, V.I.; BODNARCHUK, N.D.; KIRSANOV, A.V.

Phosphorylation of primary nitriles, Zhur. ob. khim. 33 mo.5: 1591-1596 My '63. (MIRA 16:6)

1. Institut organicheskoy khimii AN UkrSSR.
(Nitriles) (Phosphorylation)

LEVCHENKO, Ye.S.; SHEYNKMAN, I.E.; KIRSANOV, A.V.

N-dichlorophosphinylalkaniminosulfonic acid chlorides. Zhur. ob.khim. 33 no.10:3315-3323 0 163. (MIRA 16:11)

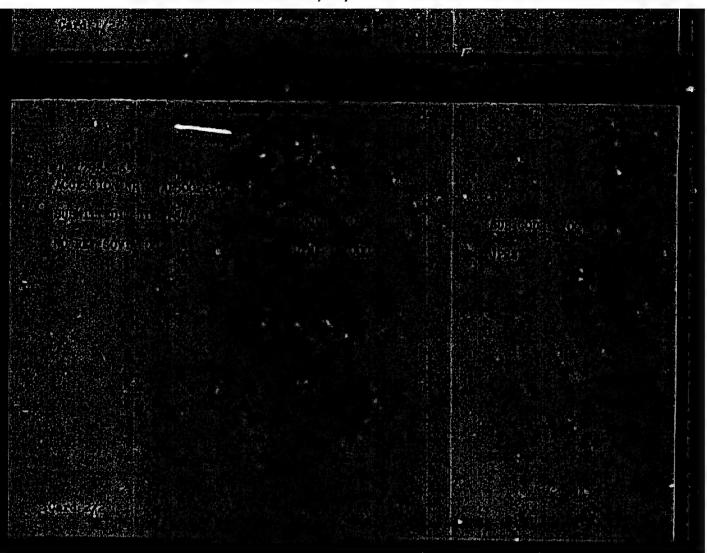
1. Institut organicheskoy khimii AN UkrSSR.

SHEVCHENKO, V.I.; TKACH, V.P.; KIRSANOV. A.V.

2,2',3,3'-Tetrabromodipropyl and 2,2'-dibromodiallyl esters of aryl-sulfonylamidophosphoric acids. Zhur.ob.khim. 34 no.2:624-627 F '64.

1. Institut organicheskoy khimii AN UkrSSR.

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LEVCHENKO, Yo.S.; KIRSANOV, A.V.

Derivatives of sulfur bisimine dioxide. Zhur.org.khim. 1 no.2:302 305 F 165. (MIRA 18:4)

1. Institut organicheskoy khimii AN UkrSSR.

ZHMUROVA, I.N.; DRACH, B.S.; KIRSANOV, A.V.

Chlorination of hydrocarbon radicals of aliphatic tricklorophosphazo compounds by phosphorus pentachloride. Ukr.khim.zhur. 31 no.2:223-224 165. (MIRA 18:4)

1. Institut organicheskoy khimit AN UkrSSR.

SHCKOL, V.A.; FEDOTOVA, L.I.; FROLOVA, A.N.; KIRSANOV, A.V.

Higher dialkyl esters of arylaulfonylamidophosphoric acids.

Zhur. ob. khim. 35 no.3:534-544 Mr '65. (MIRA 18:4)

1. Institut organicheskoy khimii AN UkrSSR.

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BODNARCHUK, N.D.; SHEVCHENKO, V.I.; KIRSANOV, A.V.

Reaction of phosphorus pentachloride with the disheryl ester of malonic acid. Zhur. ob. khim. 35 no.4:713-715 Ap. 165.

(MIRA 18:5)

1. Institut organicheskoy khimii AN UkrSSR.

SHEVCHENKO, V.I.; TKACH, V.P.; KIRSANOV, A.V.

Phenyldiallocyphosphazo sulfonyl aryls. Zhir. ob. khim. 35 no.6:992-996 Je '65. (MIRA 18:6)

1. Institut organicheskoy khimii AN UkrSSR.

ZHURAVIEVA, L.P.; BUTCVA, G.I.; KIRSANCV, A.V.

Palmitates and stesrates of trimethylphosphine oxide. Zhur.
ob. khim. 35 no.61996-998 Je 165. (MIRA 1816)

1. Institut organicheskoy khimi an UkrSSR.

ZHURAVLEVA, L.P.; GRINYUK, M.A.; KIRSANOV, A.V.

Deri-atives of phosphamic acid. Zhur. ob. khim. 35 no.6:998-1001 Je 165. (MIRA 18:6)

1. Institut organicheskoy khimii 4N Ukr6SR.

ZHMUROVA, I.N.; DRACH, B.S.; KIRSANOV, A.V.

Eydrolysis and acidolysis of trichlorophosphaze alkyle and trichlorophosphaze-A-carboxyl alkyl chlorides. Zhur. ob. khim. 35 no.6:1018-1022 Je \*65. (MIRA 18:6)

1. Institut organicheskoy khimii AN UkrGSR.

SHTEPANEK, A.S., SHEVCHENKO, V.I., KIRSANOV, A.V.

N-carbalkoxyamidophanylphan 'acids. Zhur. ob. khim. 35 no.6:1023-1025 Je 165. (MIRA 18:6)

1. Institut organicheskoy khimii AN Ukrainskoy SSR.

Derivatives of bis- \$\textit{\beta}\colon \text{alkex} vinylphorprint solds. Zhur, ch. khim. 35 no.8clv83-1487 Ag \*65. (MIRA 18:8)

1. Institut organicheskoy khimii AN Ukrūsk.

SHEVCHENKO, V.I.; PINCHUK, A.M.; KIRSAHOV, A.V.

Mixed triarylphosphazo sulfonyl aryls. Znur. cb. khim. 35 no.8;1488-1491 Ag \*65. (MiEA 18:8)

1. Institut organicheakoy khimii AN UkrSSR.

SHEVCHENKO, V.I.; KORNUTA, P.P.; KIRSANOV, A.V.

Phosphorylation of 1-cyanocarboxylic acids. Zhur. ob. khim. 35 no.9:1598-1602 S '65. (MIRA 18:10)

1. Institut organicheskoy khimii AN UkrSSR.

IVANOVA, Zh.M.; LEVCHENKO, Ye.S.; KIRSANOV, A.V.

Alkovy and aroxydihalophosphazo sulfonyl aryls. Zhur. ob. khim.
no:9:1607-1612 S \*65. (MIRA 18:10)

1. Institut organicheskoy khimii AN UkrSSR.

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the problems connected with is that first proposed by A. y of phosphazo compounds, pubpresents lists of the phosphis intended for scientists, in modern progress in organic chosphor-organic compounds. E., for which the authors give at of compounds, and an appropriate the proposed section of the phosphor-organic compounds.	iew of recent research in the inconsistencies in terminology. Mikhaelis. The book deals lished in the scientific press azo compounds that are known ndustrial workers, teachers, chemistry, especially those ach chapter deals with a different control of present the rether of the restant of the	
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	onovich; Zhmurova, Irina Nikol, Veniamin Isaakovich; Shtepan (azosoyedineniya) Kiev, Izd-voed of title: Akademiya nauk 2000 copies printed.  sphorus compound, nitrogen come introduction contains a revolute problems connected with is that first proposed by A. Ty of phosphazo compounds, public presents lists of the phosph is intended for scientists, it modern progress in organic compounds. E. for which the authors give	onovich; Zhmurova, Irina Nikolayeyna; Kirsanov, Aleksandr, Veniamin Isaakovich; Shtepanek, Alla Stanislavovna (azosoyedineniya) Kiev, Izd-vo "Naukova dumka," 1965. 283 p. end of title: Akademiya nauk Ukrainskoy SSR. Institut 2000 copies printed.  Sphorus compound, nitrogen compound, organic azo compound ne introduction contains a review of recent research in the the problems connected with inconsistencies in terminology. I is that first proposed by A. Mikhaelis. The book deals by of phosphazo compounds, published in the scientific press presents lists of the phosphazo compounds that are known is intended for scientists, industrial workers, teachers, in modern progress in organic chemistry, especially those phosphor-organic compounds. Each chapter deals with a differ of which the authors give the method of preparation, the set of compounds, and an appropriate bibliography.  [2] [3] [4] [4] [5] [6] [6] [6] [6] [6] [6] [6] [6] [6] [6

# "APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000722720006-5

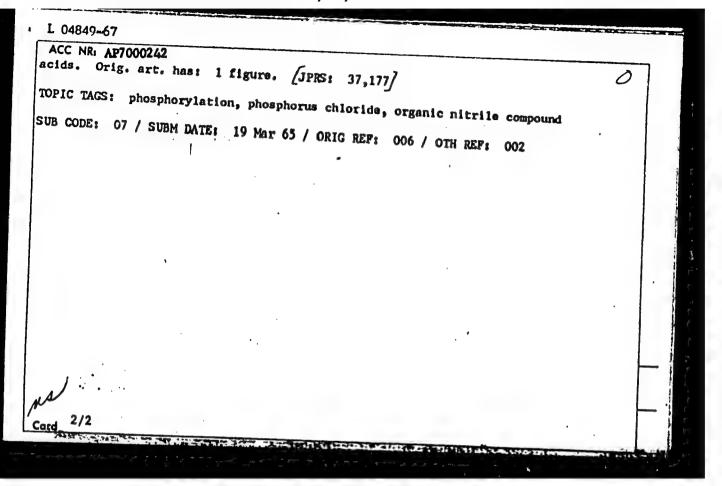
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Ch. 7. Pho	sphazoamides sphazines 238	235				
ch. 9. Pho	sphazides 258	3				
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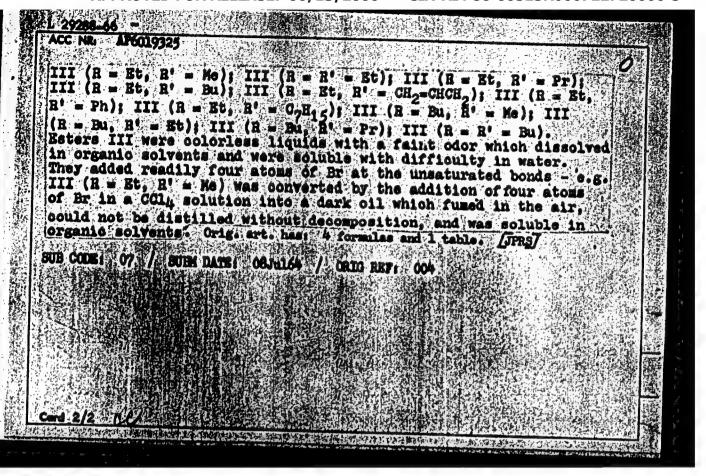
CIA-RDP86-00513R000722720006-5

L 04849-67 EWP(3)/EWT(m) RM/JW ACC NR: AP7000242 SOURCE CODE: UR/0079/66/036/004/0730/0735 25 AUTHOR: Shevchenko, V. I.; Kornuta, P. P.; Bodnarch k, N. D.; Kirsanov, A. V. ORG: Institute of Organic Chemistry, AN UkrSSR (Institut organicheskoy khimii AN UKrSSR) "Phosphorylation of Malonodinitrile by Phosphorus Pentachloride" Moscow, Zhurnal Obshchey Khimii, Vol 36. No 4. 1966, pp 730-735 Abstract: Malonodinitrile and phosphorus pentachloride, regard-less of the quantitative ratio, react at 80° and above to form acyclic trichlorophosphazo-l-chloro-, and 1,2-dichloro-2-cyano-vinyls. At 20-25°, they yield cyclic 1,1,3,5-tetrachloro-, and 1,1,3,4,5-pentachloro-1,2,6-phosphadiazines, isomeric to the acyclic phosphazo-compounds. The latter are readily converted to cyclic isomers under the action of hydrogen chloride at 20-250. The structures of the reaction products were confirmed by infrared spectra. The trichlorophosphazocyanovinyls are viscous light yellow liquids, which are readily hydrolyzed by atmospherio moisture, react vigorously with water, alcohols, and amines, undergo acidolysis, and exhibit typical properties of unsaturated compounds, such as the addition of chlorine and bromine. The phosphadiazines are colorless crystalline substances, which are slowly hydrolyzed by atmospheric moisture and react readily with water, alcohols, and organic Card 1/2 IDC: 547.461.3

## "APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000722720006-5

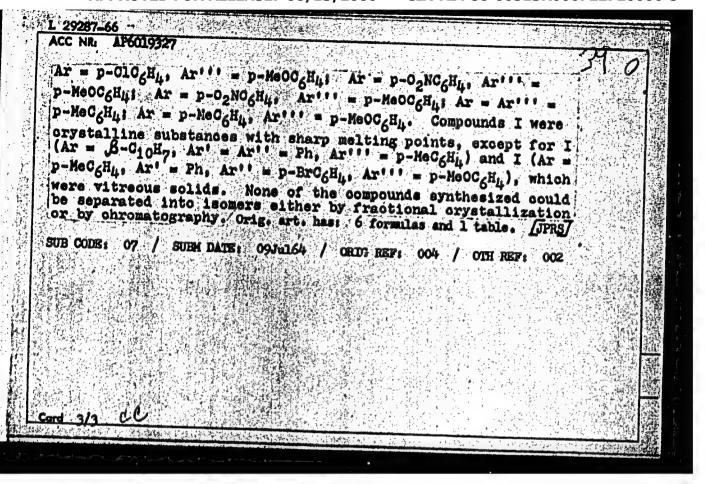


66 =BIP ()/ET(C): 10 ACC NR MAPSOL9325 BOUROE CODE 1: UR/0079/65/035/008/1483/1489 AUTHOR: Pedorova, G. K.; Kirsenov, A. V. ORGS Institute of Organic Charletor, AN Ularson (Institut organicheskoy khimis TITIE: Derivatives of bis-bets-allowy-inviphosphinic solds SOURCE; Zhurnal obshohey brimil, v. 35, no. 8, 1965, 1483-1487 TOPIC TAGS: phosphorylation, phosphorus chloride, phosphinic acid, other On phosphorylation of alkylvinyl ethers with PCl ap-ABSTRACT: plied in the molar ratio 1:4, bis-beta-alkoxyvinylphosphorus tri-ohlorides (I) formed: 4HOCH=OH2 + PCl5 (HOCH=CH)2PCl3 (I) + chlorides (I) formed: 4HOCH=CH2 + PCl5 (RUCH=CH)2PCl3 (I) +
2HOCHClMe. I(R = Bt), obtained in this manner with a yield of
70%, had a m. p. of 70=750. I (R = Bu) was an oily substance
(yield 63%) which could not be isolated in a pure state. Upon
hydrolysis of I (R = Bt, Bu) with the calculated amount of water,
the ohlorides of phosphinic acids (II) formed; (ROCH=CH)2PCl3 + H20 2HC1 + (BOCH=CH) POCI (II). They were colorless, high-boiling, oily liquids which dissolved in organic solvents. By treating acid chlorides II with alcohols R!OH, esters (ROCH=CH)2POOR! (III) were prepared. The following esters III were obtained; Card 1/2 IDC: 546.185:557.361



L 29287-66 - EMP(1)// ACC NR: AP6019327	A Jacob Barrier B	OURCE CODE: UR/0079/65/0	35/008/1DBB/13/201
AUTHOR: Shevchenko	V./I. Plnoink, A. M.		
ORG: Institute of o			. 7.4
AN Ulchser)		SSR (Institut organiches	toy bissis B
TITIE: Mixed trianyl	phosphasosulfonylaryls (		
- 3 fg	1 115		
	they khimil, v. 35, no.	8, 1965, 1488-1491	
TOPIC TAGS: organic	ynthetic process, chrom	atography, chlorinated or	
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I) were prepared	riarylphosphazosulfo	nylaryls Arso N=PAr'	Truar of the same
AMBO NOS		Feactions!	
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Arso Not A P	, Pàr'ar'   280		
1460 1711 44	r'Ar''Ar'' C12 +		
Arso <sub>2</sub> NNAC1 + 1		<b>† 1 i</b>	
Arovang + PAr	Ar! Ar! I N2 + I		
ord 1/3		The second and the second seco	
	errette er er en	WDC: 547.558.1	

# L 22287-66 ACC NR. AP6019327 Compounds prepared by reactions of different types were identical. The following compounds were synthesized by the reactions indicated. I (Ar = Ar' = Ar' = Ph, Ar = p-MeCgHi,); I (Ar = Ph, Ar'= Ar'' = Ph); I (H = Ph, Ar' = Ar'' = p-MeOCgHi, Ar''' = Ph); I (Ar = p-ClCgHi, Ar' = Ar'' = Ph, Ar'' = p-MeCgHi, Ar'' = Ph, Ar'' = p-MeCgHi, Ar' = Ar'' = p-MeOCgHi, Ar'' = Ph); I (Ar = p-ClCgHi, Ar' = Ar'' = p-MeOCgHi, Ar'' = Ph); I (Ar - p-O2NCgHi, Ar' = Ar'' = Ph, Ar'' = Ph); I (H = p-MeCgHi, Ar' = Ar'' = Ph, Ar'' = Ph); I (H = p-MeCgHi, Ar' = Ar'' = Ph, Ar'' = Ph); I (Ar - Ar' - Ar'' = Ph); I (Ar = β-C10Hi, Ar'' = Ph); I (Ar = Ph, Ar'' = Ph); I (Ar



L 34024-66 MMF(m)/EWP(1) RM
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AUT. 101: Foshchenko, N. G.; Kirsanov, A. V.
O.G. Institute of Organic C
O:G: Institute of Organic Chemistry, All Ukrss: (Institut organicheskoy khimii
TITL: Reaction of the state of
SOURCE: Reaction of phosphorous acid with iodino and alcohols
anurnal obshchoy kilmii, v. 36. no. 1 1066 ara ara
TOPIC TAGS: phosphorus acid tolder
TOPIC TAGS: phosphorus acid, iodine, alcohol, iodide, phosphoric acid, reaction
ABSTRACTOR A PLANTA A STATE OF THE STATE OF
and icdine, indicating that in contrast to the generally accepted scheme,  iodical and phosphorus, and alcohols react in 5:1:5 ratios to feme.
1 odidar land alcohols react in 5:7.5 matter accopted scheme.
iodides and phosphoric acid monohydrate. Phosphorous acid reacts with alcohols and iodides. The latter reaction can be convenient and form the purposes. Possible reaction can be convenient and form the convenient and the convenient
and alkyl iodides. The latter reaction can be convenient and for preparative and 2 tables. (IPMS: 35 coof.
purposes. Possible reaction mechanisms are outlined. Orig. art. has: 1 formula
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UDC: 516.183:547.224
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L 31808-66 EWT(m)/EWP(1) RM/WW ACC NR: AP6021676 SOURCE CODE: UR/0079/66/036/003/0467/0469 39 AUTHOR: Shevchenko, V. I.; Kukhar', V. P.; Kirsanov, A. V. ORG: Institute of Organic Chemistry. AN UkrSSR (Institut organicheskoy khimii B Ali Ukrssr) TITLE: Phosphorylation of 2-alkoxypropionitriles SOURCE: Zhurnal obshchoy khimii, v. 36, no. 3, 1966, 467-469 TOPIC TAGS: phosphorylation, organic nitrile compound, reaction rate, organic azo compound, reaction mechanism, reaction temperature ABSTRACT: 2-Alkoxypropionitriles react at 100° with phosphorus pentoxide simultaneously at the alkoxyl and nitrile groups, or only at the nitrile group. With increasing molecular weight of the alkyl radical, the reaction rate at the alkoxyl group is sharply reduced, while the reaction rate at the nitrile group remains essentially unchanged. 2-Methoxypropionitrile splits off the methoxyl group, and trichlorophosphazo-1,1,2,2,3-pentachloropropyl is formed; 2-ethoxypropionitrile gives a mixture of trichlorophosphazo-1,1,2,2,3pentachloropropyl and trichlorophosphazo-1,1,2,2-tetrachloro-3-ethoxypropyl. 2-Propoxy-, 2-butoxy-, and 2-phonoxypropionitriles react only at the nitrile group, to form trichlorophosphazo-1,1,2,2-totrachloro-3-alkoxy- and -3-phenoxypropyls, respectively. Trichlorophosphazo-1,1,2,2-tetrachloro-3-Card 1/2

propoxy_ phazo_l,; pressure split of	1,2,2,3 or dia If the a	toxy-, and - -pentachloro tilled undo lkyl halide ot yield ch	propyl wher r vacuum (0 in this ca	heatod .1 mm).	l above The a	1500 a	t atmosphe	ric	7.	/
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### "APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000722720006-5

L 31798-66 ENT(m)/EWP(1) RM ACC NR: AP6021639 SOURCE CODE: UR/0079/66/036/003/0564/0564 AUTHOR: Feshchenko, H. G.; Kirsanov, A. V. ORG: Institute of Organic Chemistry, AN UkrSSR (Institut organicheskoy khimii All Ulassat) TITL: Kethod of producing trialkylphosphine oxides directly from the alcohols, rod phosphorus, and iodino SOURCE: Zhurnal obshchoy khimii, v. 36, no. 3, 1966, 564 TOPIC And: alkylphosphine oxide, icdine, alcohol, chemical synthesis, phosphorus chomical production ABSTRACT: Trialkylphosphino oxides are produced in 85-90% yields directly from the alcohold, red phosphorus, and iodine without isolating the intermediate alkyl iodides. Trialigabi osphino oxidos wore produced from hexyl, heptyl, ectyl, nonyl, decyl, cetyl cyclob rel, and 3,5,5-trimethylheptyl alcohols, as well as from industrial mixtures of C6-Jg, Cg-C12, and other alcohols. [JPRS] SUB CODE: 07 / SUBM DATE: 140ct65 / ORIG REF: 001 LS Cord 1/1 UDC: 547.241

L 21761-66 ENT(E) ACC NR: AP6012649 SOURCE CODE: UR/0079/65/035/002/0344/0350 AUTHOR: Zhmurova, I. N.; Drach, B. S.; Kirsapor, A. V. ORG: Institute of Organic Chemistry, AN UkrSSR (Institut organicheskoy khimii AN UKTSSR) TITLE: Acid chlorides of trichlorophosphazo-trichlorophosphazo-alpha-carboxyalkyl SOURCE: Zhurnal obshchey khimii, v. 35, no. 2, 1965, 344-350 TOPIC TAGS: amino acid, chlorination, organic phosphorous compound, chloride, phosphorous chloride ABSTRACT: When, two or more moles of phosphorus pentachloride react with c -aminoacids acid chlorides of trichlorophosphaso- q-carboxylalkyle are chtained. In most cases the phosphaso-reaction is accompanied by chlorination of the alkyl group of the amino soid, where usually not less than two chlorine atoms are in the alkyl group. The mean values of stomic refractions of nitrogen for acid chlorides of trichlorophosphaso-qcarboxyalkyls and trichlorophosphascalkyls were calculated. Orig. art. has: 9 formulas and 2 tables. [JPRS] SUB CODE: 07 / SUBM DATE: 12Deq63 / ORIG REF: 007 UDC: 547.466+546.185\*131

L 23269-66 FBD/EWT(1)/EMC(k)-2/T/EWP(k)/EWA(h) IJP(c) NG

SOURCE CODE: UR/0051/66/020/003/0490/0491

AUTHOR: Kirsanov, B. P.; Selivanenko, A. S.

ORG: none

TITLE: Nonlinear losses in high-intensity laser 25,

SOURCE: Optika i spektroskopiya, v. 20, no. 3, 1366, 490-491

TOPIC TAGS: laser, nonlinear optics, Raman scattering, two photon absorption

ABSTRACT: The probability of two-photon absorpt on (W<sub>12</sub>) due to anti-Stokes Raman scattering of light from a naodymium laser by a h drogen atom as a result of the is determined practically by one term of the summations in the expression for W<sub>12</sub> (a transition involving the 2p level). Therefore, an approximate formula is obtained for the overall probability of a two-photon process using only one term of the summation for W<sub>12</sub>. The losses associated with the frequency doubling of the stimulated light due to anti-Stokes scattering within the laser medium are discussed. The gain of the anti-Stokes radiation in the ruby laser (by the stimulated emission generated in the cavity) is calculated. It is shown that frequency doubling by in existing high-intensity lasers (i.e., the beam of stimulated emission should contain both the fundamental and the second harmonic). Orig. art. has: I formula. [CS] and 1/11/16.

LEVCHENKO, Ye.S.; BERZINA, I.N.; KIRSANOV, A.V.

N-aroylareniminosulfonyl chlorides and cryl esters. Zhur. org. khim. 1 no.7:1251-1255 J1 165. (MIRA 18:11)

1. Institut organicheskoy khimii AN Ukr. SR.

KHMEL'NITSKAYA, Ye.L., prof., doktor ekon. nauk; VOLKOV, M.Ya., kand. ekon. nauk; BEL'CHUK, A.I., kand. ekon. nauk; IORDANSKAYA, E.N., ml. nauchn. sotr.; MENZHINSKIY, Ye.A.; PAVLOVA, M.A., kand. ekon. nauk; VASIL'KOV, N.P., kand. ekon. nauk; ARDAYEV, G.B., kand. ekon. nauk; VAL'KOV, V.A., kand. ekon. nauk; TIMASHKOVA, O.K., kand. ekon. nauk; ANDREYEV, Yu.K., ml. nauchn. sotr.; PUSHKIN, A.A., ml. nauchn. sotr.; MAKSIMOVA, M.M., kand. ekon. nauk; KIRSANOV, A.V., kand. ekon. nauk; SHEBANOV, A.N., ml. nauchn. sotr.

[Changes in the economic structure of the countries of Western Europe] Izmenemiia v ekonomicheskoi strukture stran Zapadnoi Evropy. Noskva, Nauka, 1965. 433 p. (MIRA 18:9)

1. Akademiya nauk SSSR. Institut mirovoy ekonomiki i mezhdu-narodnykh otnosheniy.

SHEVCHENKO, V.I.; TKACH, V.P.; KIRSANOV, A.V.

Isomerization of trialkoxyphosphazo sulfonylaryls. Zhur. ob. khim. 35 no.7:1224-1227 Jl \*65. (MIRA 18:8)

1. Institut organicheskoy khimii AN UkrSSR.

DERKACH, Grigoriy J'larionovich; ZHMUROVA, Irina Nikolayevna;
KIRSANOV, Aleksandr Vasil'yevich; SHEVCHENKO. Vanismin
Isaakovich; SHODANEK. Alla Stanislavovni POKROVSKAYA,

[Phosphazo compounds] Fosfazosoedineniia. Kiev, Naukova dumka, 1965. 283 p. (MIRA 18:8)

ACC NR. AP6015921

SOURCE CODE: UR/0286/65/000/015/0031/0031

DEVERTOR: Kirsanov, Ar. V.I. Shevchenko, V. I.I. Tench, V. P.

ORO: none

TITLE: Method for obtaining triallyloxychosphasosulfonylaryls—Certificate So.
173227, Class C 07f

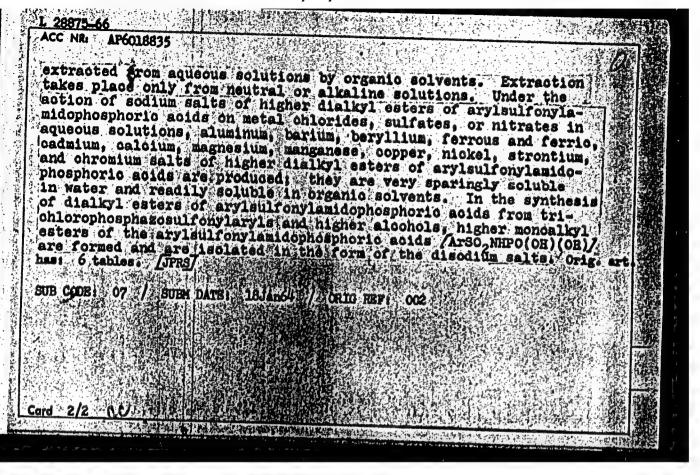
SOURCE: Ryulleten' isobretenty i tovarnyth snakov, no. 15, 1965, 31

TOPIC TAGS: polyser, organic phosphorus compound, organic sulfur compound, organic nitrogen compound

ARSTRACT: The method for obtaining triallyloxyphosphasosulfonylaryls distinguished by the fart that trichlorophosphasosulfonylaryls are treated with sodium allylate or additionally with triallyl phosphits. The use of triallyloxyphosphasosulfonylaryls sodiumolloxande with triallyl phosphits. The use of triallyloxyphosphasosulfonylaryls.

SUB CODE: 07 / SUBM DATE: 27DeoGl

ENP(1)/ENT(n)/ Ru AP6018834 BOURCE CODE: UR/0079/65/035/003/0534/0504 AUTHOR Shokol, V. A.; Fedotova, L. I.; Prolova, A. N.; Kirsonov, A. ORG: Institute of Organic Chemistry, AN Worden (Institut organicheskoy khizdi AN Ulrega TITIE: Higher diallyl esters of anylaulfonylamidophosphoric soids SOURCE: Zhurnal obshohey khimit, v. 35, no. 3, 1965, 534-544 TOPIC TAGS: organic synthetic process, ester, phosphoric acid, organic sulfur compound, organic nitrogen compound, organic salt Dialkyl esters of arylaulfonylamidophosphoric acids with higher aldyl radicals were synthesized and investigated as complex formers and extraction reagents for various metals. Dialkyl esters of arylsulfonylamidophosphoric acids, possessing the properties of monobasic acids, were synthesized by the action of trichlorophosphazosulfonylaryis on higher aliphatic alcohols or by the action of dichlorides of arylsulfonylamidophosphorio scide on higher sodium alcoholates. The solubility of the sodium salts of higher alkyl esters of arylsulfonylamidophosphoric acids in water decreases, while that in organic solvents increases with increasing molecular weight of the alkyls. Sodium salts of the higher dialkyl esters of arylaulfonylamidophosphorio acids are 546,185,547,541,521,1



1 25595-66 ENT(a)/ENP(1) WW/RM ACC NR JP6016695 SOURCE CODE: UR/0079/65/035/009/1598/1602 AUTHOR: Shevohenko, V. I.; Kormita, P. P.; Kirsanov, A. V. ORG: Institute of Organic Chemistry, AN UlcrSSR (Institut organichesky) Mitmit TITIE: Phosphorylation of 1-cyanocarboxylic acids SOURCE: Zhurnal obshohey khindi, v. 35, no. 9, 1965, 1598-1602 TOPIC TAGS: phosphorylation, phosphorus chloride, organic mitrile compound chlorination, IR spectrum, chloride, organic aso compound ABSTRACT: The reaction of phosphorus pentachloride with 1-cyanocarboxylic acids of the AlkCH(CN)COOH type was studied in an effort to expand the limits of application of the phosphorylation of nitriles. The direction of the reaction and yield of the final products depend on the quantitative ratio of the reagents and on the volume of the alkyl radical; the reaction direction is dependent to a lesser degree on the temperature. At a 1:1 molar ratio of 1-cyanocarboxylic acid and phosphorus pentachloride, the acid chloride is formed readily and rapidly. The acid chlorides formed react with phosphorus pentachloride as typical secondary nitriles, forming a mixture of phosphorylation products and the chlorides of 1-chlory-1-cyano carboxylic acids. At a 1:2 molar ratio of the 1-cyanocarboxylic acid and 546.185+547.468

### L 25595-66

## ACC NR. #26016695

phosphorus pentachloride, the phosphorylation products obtained are tri-chlorophosphaso-1-chloro-2-chlorocarboxy-2-alkylvinyls, phile at a 1:3 ratio, the phosphorylation products are trichlorophosphaso-1,1/2-trichloro-2-chlorocarboxyalkyls. As the molecular weight of the alkyl radical is increased. the yield of the phosphorylation products is sharply reduced, while the yield of the chlorination products increases (from 15% for 1-cyanopropionic acid to 62% for 1-cyanoisovaleric acid). The introduction of halogen atoms into the methyl group of isobutyronitrile sharply increases the yield of phosphoryplation products, from 40% for isobutyronitrile to 80% for 2-chloroisobutyronitrile. The chlorocarboxy group exerts the same influence as the chloromethyl group. Trichlorophosphazo-1-chloro-2-chlorocarboxy-2-methylvinyl was the only unsaturated phosphaso compound isolated in the individual state; the other unsaturated phosphaso compounds were converted without isolation to trichlorophosphazo-1,1,2-trichloro-2-chlorocarbowalkyls by the action of phosphorus pentachloride. Trichlorophosphazo-1,1,2-trichloro-2-chlorocarboxyalkyls react with acetic acid, forming dichlorides of N-dichlorophosphonylmonoiminoallylchloromalonic acids. The infrared spectra of the reaction products are discussed. Orig. art. has: 3 tables. [JPRS]

SUB CODE: 07 / SUBM DATE: 15Aug64 / ORIG REF: 006

Card 2/2 6

ACC NR: AP6018498  AUTHOR: Shevchenko, V. I.; Kormuta, P. P.; Miranov, A. V.  ORG: Institute of Organic Chemistry: AK Unissa (Institut organicheskoy khimii AH Unissa)  TITLE: Phosphorylation of secondary nitriles  SOURCE: Zhurnal obshchey khimii, V. 35, no. 11, 1965, 1970-1973  TOPIC TAGS: phosphorylation, organic nitrile compound, phosphorus chloride,  phosphoric acid, amine  ABSTRACT; Isobutyronitrile was used as a model for a detailed study of the reaction of secondary nitriles with phosphorus pentachloride. Secondary nitriles  of the CH3(Alk)URCH type react with phosphorus pentachloride at 200 to form a take place at 0-50. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride; only 1-chloroisobuty-presence of excess phosphorus pentachloride; only 1-chloroisobutyronitrile and trichlorophosphase-1-chloro-2-methylvinyl are formed; in the phosphosphase-1, 2-trichloro-2-methylvinyl are formed. If the reaction is phosphase-1, 1, 2-trichloro-2-methylvinyl are formed. If the reaction is
AUTHOR: Shevchenko, Y. I.; Kornita, P. P.; Rivanov, A. V.  Oid: Institute of Organic Chemistry: AN United (Institut organicheskoy khimii AN United Source: Phosphorylation of secondary nitriles  SOURCE: Zhurnal obshohey khimii, v. 35, no. 11, 1965, 1970-1973  TOPIC TAGS: phosphorylation, organic nitrile compound, phosphorus chloride, phosphoric acid, amine  ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the of the CH <sub>2</sub> (Alk)CRCH type react with phosphorus pentachloride. Secondary nitriles and phosphorylation products. The reaction does not take place at 0-50. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-phosphorus pentachloride, only 1-chloroisobuty-phosphorus pentachloride, only 1-c
TITLE: Phosphorylation of secondary nitriles  SOURCE: Zhurnal obshohey khimii, v. 35, no. 11, 1965, 1970-1973  TOPIC TAGS: phosphorylation, organic nitrile compound, phosphorus chloride, phosphoric acid, amine  ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the of the CH3(Alk)CRCM type react with phosphorus pentachloride. Becomdary nitriles mixture of 1-chloronitriles and phosphorylation products. The reaction does not romitrile and trichlorophosphano-1-chloro-2,2-dimethylvinyl are formed; in the trichlorophosphane-1,1,2-trichloro-2-methylvinyl are formed; in the conducted and reaction of access phosphorus pentachloride, only 1-chloroisobuty-onitrile and trichlorophosphane-1,1,2-trichloro-2-methylvinyl are formed; in the
TITLE: Phosphorylation of secondary nitriles  SOURCE: Zhurnal obshohey khimii, v. 35, no. 11, 1965, 1970-1973  TOPIC TAGS: phosphorylation, organic nitrile compound, phosphorus chloride, phosphoric acid, amine  ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the of the CH3(Alk)CRCM type react with phosphorus pentachloride. Becomdary nitriles mixture of 1-chloronitriles and phosphorylation products. The reaction does not romitrile and trichlorophosphano-1-chloro-2,2-dimethylvinyl are formed; in the trichlorophosphane-1,1,2-trichloro-2-methylvinyl are formed; in the conducted and reaction of access phosphorus pentachloride, only 1-chloroisobuty-onitrile and trichlorophosphane-1,1,2-trichloro-2-methylvinyl are formed; in the
TOPIC TAGS: phosphorylation, organic nitrile compound, phosphorus chloride, phosphoric acid, amine ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the of the CH <sub>3</sub> (Alk)CRCM type react with phosphorus pentachloride. Secondary nitriles mixture of 1-chloronitriles and phosphorus pentachloride at 200 to form a take place at 0.50. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobutyronitrile and trichlorophosphage-1,1,2-trichloro-2-methylyinyl are formed; in the
phosphoric acid, amine  ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the of the CH <sub>2</sub> (Alk)CHCH type react with phosphorus pentachloride. Secondary nitriles with phosphorus pentachloride at 200 to form a take place at 0.50. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobutyronitrile and trichlorophosphese-1,1,2-trichloro-2-methyly(ny) are formed; in the conducted areas and phosphorus pentachloride, only 1-chloroisobutyronitrile and trichlorophosphese-1,1,2-trichloro-2-methyly(ny) are formed; in the
ABSTRACT: Isobutyronitrile was used as a model for a detailed study of the reaction of secondary nitriles with phosphorus pentachloride. Secondary nitriles of the CH <sub>1</sub> (Alk)CRCH type react with phosphorus pentachloride at 200 to form a mixture of 1-chloronitriles and phosphorylation products. The reaction does not romitrile and trichlorophosphaso-1-chloro-2,2-dimethylvinyl are formed; in the trichlorophosphase-1,1,2-trichloro-2-methylvinyl are formed; in the conducted area.
reaction of secondary nitriles with phosphorus pentachloride. Secondary nitriles of the CH <sub>2</sub> (Alk)CHCH type react with phosphorus pentachloride. Secondary nitriles wixture of 1-chloronitriles and phosphorus pentachloride at 20° to form a take place at 0-5°. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobutyronitrile and conducted and area.
mixture of 1-chloronitriles and phosphorus pentachloride at 200 to form a take place at 0.50. In the presence of excess nitrile, only 1-chloroisobuty-presence of excess phosphorus pentachloride, only 1-chloroisobuty-trichlorophospheze-1,1,2-trichloro-2-methylyinyl are formed; in the conducted areas and conducted areas phosphorus pentachloride, only 1-chloroisobutyronitrile and conducted areas areas are conducted areas are conducted areas areas are conducted area
take place at 0.50. In the presence of excess nitrile, only 1-chloroisobuty- presence of excess phosphorus pentachloride, only 1-chloroisobuty- trichloropiosphere-1,1,2-trichloro-2-methylyinyl are formed; in the
ronitrile and trichlorophospheso-1-chloro-2,2-dimethylvinyl are formed; in the trichlorophospheso-1-chloroide, only 1-chloroisobuty-trichlorophosphese-1,1,2-trichloro-2-methylvinyl are formed; in the conducted and area.
presence of excess phosphorus pentechloride, only 1-chloroisobutyronitrile and
trichloropiospheze-1,1,2-trichloro-2-methylytivil
conducted at most grant of the conduction of the
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conducted at 700 or above, only 1-chloroisobutyronitrile is formed. If the reaction is phosphago-1,1,2-trichlorosikyle of the CH <sub>3</sub> (Alk)COICI_N=POI <sub>3</sub> type are colorless
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are readily decomposed by atmospheric moisture and read vigorously with amines, and phosphoric soid, and decompose at 150/2000 into phosphorus
and phosphoric acid, and decompose at 150,200° into   hosphorus pentachloride and
1-chloronitriles. The thermal stability of phosphase compounds of this type
,如 <b>如此,一个人也以上</b> 对人,不是可能是一个,我们就能够被抗议的,就是我们的,我们就是一个人,我们就能够是一个人,我们就能够是一个人,我们就能够是一个人,我们就能够
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7768-66 - SIP(4)/BIT(B) ACC NRI AP6018499 SOURCE CODE: UR/0079/65/035/011/1974/1976 AUTHOR: Ivanova, Zh. M.; Kirsanov, A. V. ORG: Institute of Organic Chemistry, AN UkrSSR(Institut organicheskoy khimii AN UkrSSR TITLE: Fluorochloride of phenylphosphonic acid and fluorides of monoamides of SOURCE: Zhurnal obshchey khimii, v. 35, no. 11, 1965, 1974-1976 TOPIC TAGS: fluorinated organic compound, phosphorus chloride, primary aromatic amine, secondary amine, organic synthetic process, organic phosphorus compound, ABSTRACT: The reaction of fluorides of monoesters of phenylphosphonic scid with phosphorus pentschloride results in the formation of the fluorochloride of phenylphosphonic acid, which reacts with primary and secondary suines to give fluorides of monoemides of phenylphosphonic scid. A simpler method was developed for the synthesis of fluorides of monoesters of phenylphosphonic acid, the starting meterials for the production of the fluorochloride of phenylphosphonic acid: the dichloride of phenylphosphonic acid is successively treated with alcohol and potessium bifluoride without isolating the chlorides of monoesters of phenylphosphonic acid. Orig. art. has: 4 formulas. [JPRS] SUB CODE: 07/ SUBM DATE: 090ct64 ORIG REF: 003 / OTH REF: 003 Cord - 1/1 1/1/ UDC: 547.241

EMP(4) ACC NR. AP6018502 SOURCE CODE: UR/0079/65/035/011/1984/1988 AUTHOR: Pedorova, G. K. | Shaturskiy, Ys. P. Kirsanov, A. V. OHG: Institute of Organic Chemistry, AN UkrSSR (Institut organicheskoy khimii AN UKTSSRT TITLE: Derivatives of styry1-2-chlorostyrylphosphinic and bis-phenylacetylenyl-SOURCE: Zhurnal obshchey khimii, vo. 35, no. 11, 1965, 1984-1988 TOPIC TAGS: phosphorylation; ester, phenol, amine, chlorinated organic compound, organic phosphorus compound, hydrolysis, nonmetallic organic derivative ABSTRACT: Phenylacetyleise to phosphorylated by styryltetrachlorophosphorus, forming styryl-2-chlorostyryltrichlorophosphorus, Styryl-2-chlorostyryltrichlorophosphorus is hydrolysed to the corresponding soid, and reacts with sulfur dioxide to give the chloride of styry1-2-chlorostyrylphosphinic soid. Treatment of styryl-2-chlorostyrylphosphinic and bis-2-chlorostyrylphosphinis Treatment or styryl-2-chlorostyrylphosphinic and bis-2-chlorostyrylphosphinic acids with alcoholic potassium hydroxide results in the formation of styrylphosphinic and bis-phenylacetylenylphosphinic acids. Under the action of phenols and srouatic amines, the chlorides of styryl-2-chlorostyrylphosphinic and bis-phenylacetylenylphosphinic acids are converted to the corresponding esters and smilides yields, melting points, crystal type, and analytic data are given for all the reaction products. Orig. art. SUB CODE: 07 / SUBH DATE: 30Nov64 ORIG REFI OOL/

### "APPROVED FOR RELEASE: 06/13/2000

### CIA-RDP86-00513R000722720006-5

Description of the process of crossing over the Belation to the Change in Doses of Outer and Inner Pactors in the Broophila Melanogaster," B. A. Kirsanov, h pp

"Dok Alad Haak 8888, Nova Ser" Vol LVIII, No 3

Discusses variations in the process of crossing over them acted upon by: 1) various doses of an outside factor, in this case I-rays; and 2) variations in desse of an internal factor—changing the amount of invariation in menhanologous chromocomes. Also studied changes in the frequency of fission in special relation to various doses of I-rays, additited by Academicianal. I. Beaut. Section 1, 1987.

# KRONGAUZ, A.H.; BRODOVSKIY, H.P.; SHWYKOLOVICH, Yu. V.; KIRSANOV, B.A.

Stand for measuring external gamma irradiation in radioactive preparations. Vest. rent. i rad. 33 no.6:64-69 N-0 '58. (MIRA 12:1)

1. Is dosimetricheskogo otdela (sav. - dota. A.N. Krongaus) Gosudarstvennogo instituta rentgenologii i radiologii (dir. - dota. I.G. Legunova) Ministerstva sdravookhraneniya RSFSR. (RADIOLOGY, appar. & instruments

stand for measurement of external gamma rays of isotopeenclosing containers (Rus))

L 10841-66 SWT(1)/SWT(a)/EWP(J)/EWA(G)/ T IUP(c) AT/GS/RM ACC NR AT5023436 SQURCE CODE ( UR/0000/65/000/000/0110/011 AUTHOR: Krongelin V. A. Verilly Kireanov, B. ORG ! none TITIMIT Investigation of the mechanism of intermolecular energy transfer in organic solutions. Effect of diffusion SOURCE: Simposium po elementarnym proteessem khimii vysokikh energiy. Hoscor, 1983 ile intermy/e professory khis! (ysokikh energiy (Elementary processes of the chemis try of high energies) trudy simpos(uma: Moscow; 1965, 110-113 TOPIC TAGS: excited state; particle interection, molecular interaction, particle collision, luminescence ABSTRACT: The transfer of excitation energy between beasans and toluene 2 3-diphesing loxazole and 2 5-diphesional and leopropyldiphenyl/and cyclohexans/was studied the dependence of the relative intensity () of luminescence of a diphenyloxazole solution (0.005 moles/i) in isopropyldiphenylcyclohexans mixture upon the reciprocal viscosity of the solvent is shown in liquid 15 For all three systems the experiment ally determined sate constants of energy transfer / a me lower than those calculates APPROVED FOR KELEASE: 00213/2000 CIA-RDP86-00513R000722720005 Experience in using reinforced concrete poles in a contact system.

Transp.strei.5 ne.8:14-15 0 '55. (MIRA 9:1)

1.Zamestitel' nachal'nika 1-ge uchastka energesnabsheniya Oktyabr'skoy doregi. (Electric lines--Poles)

